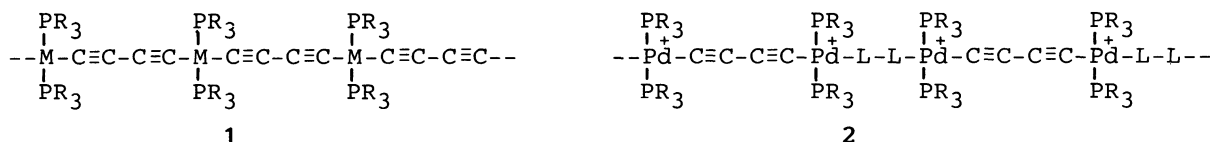


Synthesis and Characterization of a New Class of Organo-polynuclear Complexes Containing Cationic Palladium Atoms

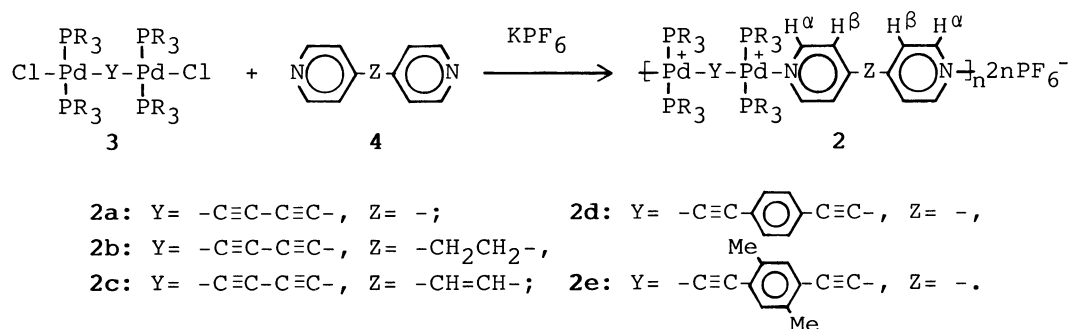
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Cationic organopalladium complexes having an extended structure are described as a new class of organo-polynuclear complexes containing ionic transition metals with regular arrangement in the main chain.

Organometallic polymers have drawn much attention in terms of their unique physical properties and catalyses.¹⁾ However, there are few examples of organometallic polymers containing transition metals in the main chain.^{2,3)} We previously synthesized transition metal-poly(yn) polymers (1) which have a backbone composed of transition metals and conjugated acetylenic linkages.³⁾ Here we wish to report the synthesis and properties of a new type of polynuclear complexes (2) which have unique backbones composed of cationic palladium atoms with regular arrangement and alternate linkages of acetylene and bipyridyl.



Thus, μ -butadiynediyl-bis[trans-chlorobis(tri-*n*-butylphosphine)-palladium](3a)⁴⁾ was reacted with 4,4'-bipyridyl (molar ratio, 1/1) in the presence of excess KPF₆ in acetone to give an ionic organo-polynuclear complex (2a, Y = --C≡C--C≡C--, Z = -, R = *n*-Bu) in 88% yield as yellow solids; decomp. at 104-108 °C, [η] = 0.97 dL/g in dichloromethane at 25 °C. The average molecular weight of 2a was tentatively obtained to be ca. 10000 by VPO and NMR methods (vide infra), which corresponds to n = ca. 6, and hence 2a has about twelve palladium atoms in a molecule. The polynuclear complex is stable in air and soluble in polar organic solvents such as dichloromethane and acetone. The structure was characterized by spectral methods. The ³¹P{¹H} NMR spectrum showed a single peak due to the PBu₃ ligands at 16.25 ppm down field from an external PPh₃ reference, and the resonance due to the PF₆ appeared at 138.84 ppm up field from the reference (J_{P-F} = 711 Hz), indicating that the terminal groups of 2a are not chlorine atoms but bipyridyl ligands. The ¹H-NMR (360 MHz) exhibited resonances of H^α at δ 9.06 and H^β at 8.27 along with weak resonances at δ 8.79 and 7.82. The latter pair may be assigned to the hydrogens bonded to uncoordinated pyridyl moieties of the terminal bipyridyl ligands and used for molecular weight estimation. These spectral data are



consistent with a regular structure of **2a**. As compared with the neutral polymers (**1**), the stretching frequencies of $\text{C}\equiv\text{C}$ in the IR spectra of **2** shift to higher frequencies and the phosphorus chemical shifts in ^{31}P -NMR spectra to lower magnetic fields due to the cationic character of the palladium atoms. Polynuclear complex **2a** showed electric conductivities in polar solvents; $596 \mu\Omega^{-1}\text{cm}^{-1}$ in acetone, 754 in nitromethane.

Similarly several analogues (**2b-e**) were prepared from the reaction between selected pairs of the dihalides (**3**) and bipyridyl derivatives (**4**). They may provide a new class of ionic organo-polynuclear complexes having transition metals in the main chain, though some metallocene polymers containing ferricinium⁵⁾ and cobalticinium⁶⁾ species are known. The concentrated solutions of the polynuclear complexes such as **2a** and **2d** in dichloromethane exhibited formation of nematic liquid crystals indicating that the polynuclear complexes have an extended linear structure.⁷⁾ The high resolution $^{31}\text{P}\{^1\text{H}\}$ NMR study⁷⁾ of the lyotropic liquid crystals revealed that polynuclear complex **2a** aligned its main chain in the direction parallel to the applied magnetic field (23.5 kG; $1 \text{ G} = 10^{-4} \text{ T}$).⁸⁾

This work was supported by Grant-in-Aid for Scientific Research from the Ministry of Education, Science and Culture.

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(Received August 9, 1988)